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Molar gas volume = 24.0 dm3 mol -1 at room temperature and pressure, RTP Avogadro constant, N A =  $6.02 \times 1023$  mol -1 Specific heat capacity of water, c = 4.18 J g -1 K -1 Ionic product of water, K w =  $1.00 \times 10-14$  mol 2 dm -6 at 298 K 1 tonne = 106 g Arrhenius equation: k = Ae-E a/RT or ln k = -E a/RT + ln A ...

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